TOTAL SYNTHESIS OF TYLONOLIDE, THE AGLYCONE OF THE 16-MEMBERED RING MACROLIDE TYLOSIN, FROM D-GLUCOSE. SELECTIVE APPLICATION OF MPM AND DMPM PROTECTING GROUPS FOR HYDROXY FUNCTIONS¹

Tatsuyoshi Tanaka, Yuji Oikawa, Tatsuo Hamada, and Osamu Yonemitsu* Faculty of Pharmaceutical Sciences, Hokkaido University, Sapporo 060, Japan

Summary Segments i (Cll-C17) and ii (Cl-ClO), synthesized from D-glucose by employing some stereoselective reactions and benzyl-type protecting groups, were esterified and cyclized to the 16-membered enone, which was readily converted to tylonolide, the aglycone of tylosin.

Tylosin is a typical 16-membered ring macrolide antibiotic that is important not only as a therapeutic agent but also as a target molecule in modern synthetic organic chemistry. In connection with our synthetic study of macrolide and polyether antibiotics from D-glucose by a common methodology consisting of stereocontrolled reactions and the MPM (4-methoxybenzyl) protection,² a highly stereoselective and efficient synthesis of methynolide, the aglycone of the 12-membered ring macrolide methymycin, was recently achieved.¹ This methodology has been successfully applied to the total synthesis of tylonolide (1) , 3 the aglycone of tylosin. The general synthetic strategy consists of condensation and cyclization of segments i (2) and ii (3), and selectivity between DMPM (3,4_dimethoxybenzyl) and MPM protecting groups in the deprotection with DDQ^{2c,e} is crucial in the final stage of the synthesis of 1.

Segment i (2) was easily synthesized from glucose via the known alcohol (4) .⁴ The alcohol (4) was first protected with MPM, 2a and the 5,6-side chain⁵ was reduced to give 6, which was also synthesized from glucose via 5.¹ After hydrolysis of the acetonide group of 6, the

(A) 1) NaH, MPMC1, (93%); 2) $2\%-\text{H}_2\text{SO}_4$, MeOH, rt (96%); 3) TsC1, NaI(80%); 4) H₂, Pd-C (100%). (B) 1) $\texttt{Ph}_3\texttt{P=CH}_2$ (82%); 2) \texttt{BH}_3 , THF; $\texttt{H}_2\texttt{O}_2$, NaOH (55%); 3) NaH, MPMCl (80%). (C) 1) 4.5N-HCl, THF, 50 C (72%); 2) $\text{Ca}(\text{BH}_4)_2$, EtOH (86%); 3) NaIO₄, MeOH-H₂O (95%). (D) Ph₃P=C(Me)CO₂Et, C₆H₆, 60°C (91%); 2) LiAlH₄, THF (97%); 3) MnO₂, CH₂C1₂ (96%).

resulting hemiacetal group was reduced with Ca(BH $_{\rm A})_{\rm 2}$, $^{\rm 0}$ f $_{\rm 6}$ 7, $[\alpha]_{\rm D}$ ¹⁵ + 6.9°.′ T followed by treatment with \texttt{NaIO}_Λ to give The Wittig reaction of 7 with a stable ylid gave the expected E-ester, which was reduced with LiAlH₄, followed by oxidation with MnO₂ leading easily to the segment i (2) in excellent yield.

Segment ii (3) has a more complex structure. However, since the stereochemical configurations at C-4 \sim C-8 are identical to those at C-2 \sim C-6 of methynolide, the methodology employed in the synthesis of the segment ii of methynolide^l can be applied to the synthesis of 3. Compound 8^8 easily available from glucose was converted to 9 via a series of six conventional reactions. Catalytic reduction of 9 with various catalysts easily gave a dihydro compound, but this reduction was not practically useful because of poor stereoselectivity $(2.5 \times 3.0 : 1)$. Another route via hydroboration proceeded with an excellent stereoselectivity (16 : 1) to give 10, $[\alpha]_n^{20}$ + 68°, which has three consecutive chiral centers corresponding to the C-4 \sim C-6 of 1. The chiral center at the C-8was constructed via the aldehyde (11) and the acetal (12) **to** give 13, $\left[\alpha\right]_D^{16}$ + 97°, ⁹ by a method similar to that described in the construction of the C-6 chiral center of methynolide. $¹$ </sup>

14: $R^1 = CH_2OH$, $R^2 = TBDMS$ 18: R=DMPM 15: R^1 =CHO, R^2 =TBDMS

 $0 - R^2$

17: $R^1 = R^2 = COC_6H_4 - NO_2 - p$
19: $R^1 = H_1$. $R^2 = DMPM$

(E) 1) LiA1H₄; 2) NaH, BnCl (92%, overall); 3) 2%- $H₂SO₄$, MeOH, rt (92%); 4) TBDMSC1, imidazole, CH₂C1₂ (95%); 5) DMSO, (COC1)₂, NEt₃, CH₂C1₂, - 60 : $R^2 = R^2 = CO_6H_4 - NO_2^-P$ $C^2 (95z); 6) Ph_3P = CH_2$, $THF^2 (92z). (F) 1) BH_3$, 19: R^1 =H, R^2 =DMPM $\begin{array}{ccc} \text{THF}_\bullet - \text{30}^\circ\text{C}; & \text{H}_2\text{O}_2, \text{ NaOH (373)}; & 2) \text{ TsC1, pyridine;} \ \text{3) LiAlH}_4, \text{ Et}_2\text{O (88\%, overall).} \end{array}$ (G) 1) NaH, MPMC1 (92%); 2) 4N- $\bar{\texttt{H}}_2$ SO $_4$, dioxane, 60°C (68%); 3) NaIO $_4$,

MeOH-H₂O (95%). (H) 1) (MeO)₂P(O)CH(Me)CO₂Me, NaH, THF, - 80~0°C; 2) K₂CO₃, MeOH (68%,
overall); 3) DIBAH, toluene, - 78°C; 4) CSA, i-PrOH (93%, overall). (I) 1) H₂, Raney Ni, EtOH (81%); 2) TBDMSC1, imidazo1e, CH₂C1₂ (95%); 3) H₂, 1 DMSO, (COC1) $_2$, NEt $_3$, CH $_2$ C1 $_2$, Pd-C (96.5%) or DDQ, CH₂C1₂-H₂O (94%); 4) - 60°C (100%). (J) 1) CH₂=CHCH₂I, CrC1₂, O°C (78%); 2) n-Bu₄NF, THF (98%); 3) NaH, DMPMC1 (70%). (K) 1) 1N-HC1, THF, 50°C; 2) Ca(BH $_{\Delta}$) $_{2}$, EtOH (90%, overall); 3) Me $_2$ C(OMe) $_2$, CSA, C $_6$ H $_6$ (86%). (L) 1) DMSO, (COC1) $_2$, NEt $_3$, CH $_2$ C1 $_2$, - 60 °C; 2) MeP(O)(OMe) $_2$, n-BuLi, THF, - 78°C; 3) PDC, DMF (60%, overall); 4) KMnO $_{\Delta}$, NaIO $_{\Delta}$, Me₂CO-H₂O (83%).

The final chiral center at the C-3 was introduced by the Cram addition 10 of a Grignard reagent to the aldehyde (15). The primary alcohol of 13 was protected with a TBDMS (t-butyldimethylsilyl) group, and the MPM group was removed to give 14, which was readily oxidized to the aldehyde (15). When 15 was treated with $CH_2=CHCH_2MgBr$ at - 90°C, a stereo-mixture (6.7 : 1) mainly containing the Cram adduct $(16; 90\%)$ was isolated. A better stereoselectivity $(10 : 1)$ was obtained by the reaction with CH₂=CHCH₂I in the presence of CrC1₂.¹¹ The structure of 16 was confirmed after conversion to 17, in which NOE between the methyl group* and H_a and H_b were observed to be 13 and 14%, respectively. Thus the introduction of all chiral centers of the segment ii (3) was completed. The silyl protection of 16 was replaced by the DMPM group $^{2\mathsf{c}}$ to give 18, which was converted to 3 via 19 in almost the same manner as described in the preceding paper. Thus 19 readily prepared from 18 was oxidized and condensed with a phosphonate, 12 followed by immediate PDC (pydridinium dichromate) oxidation 13 to give the ketophosphonate. Finally, the terminal olefin was oxidized under Lemieux-von-Rudloff conditions¹⁴ to give the segment ii (3), $[\alpha]_n^{26.5}$ - 5.6°.

Esterification between 2 and 3 proceeded by the usual treatment with DCC, but the Yamaguchi method 15 gave a better result. The resulting ester (20), [$\alpha]_{\rm D}^{-26}$ + 2.8°, was subjected to the Aristoff-Nicolaou cyclization^{3c, 16} at 100°C for 12 hr to give the 16-membered enone (21), [α] n^{26} + 6.2°, in acceptable yield. Four steps conversion of 21 into tylonolide (1) as follows. Selective deprotection of the DMPM group^{2c} with DDQ at 0° C gave the primary alcohol, which was converted to the aldehyde by the PDC oxidation, 13 and then acid-treatment to remove the isopropylidene protection gave the hemiacetal compound. Finally, the MPM protection was removed by re-treatment with DDQ under usual conditions^{2a} to give tylonolide (1; mp 103°C), which was completely identical in its spectral data (IR, NMR, Mass) with 1 derived from natural ty losin.^{17,18} In this total synthesis, stereoselectivities for the construction of new chiral centers at the C-3, C-4, C-6, C-8, and C-14 were 91, 94, 100, 87, and 100%, respectively.¹⁹

(M) 1) 2,4,6-Cl₃C₆H₂COCl, NEt₃, THF, DMAP, C₆H₆, rt, 1 h (66%). (N) K₂CO₃ (6eq), 18-crown-6 (12eq), toluene, 100°C, 12 h (59%). (0) 1) DDQ (1.2eq), CH₂C1₂-H₂0, O°C (75%); 2) PDC, CH₂C1. (94%); 3) IN-HCl, THF, rt (90%); 4) DDQ, $CH_2Cl_2-H_2O$, rt (70%).

Aknowledgment. We are grateful to Professor M. Yamaguchi and Dr. J. Inanaga, Kyushu university, for their kind gift of the precious reagent, 2,4,6-trichlorobenzoyl chloride.

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18. The synthetic tylonolide (1) is a 3.3:1 isomeric (equilibrium) mixture with respect to the hemiacetal position. When a CDCl₃ solution of 1 derived from the natural tylosin by Grieco's method C^{17a} [6:1 mixture, mp 155-156°C (1it.^{17a} mp 102.5-103.5°C or 157.5-158.5°C)] was allowed to stand in a NMR tube, isomerization probably with a trace of acid occurred to give a 3:1 mixture.

19. 1: δ (500 MHz): 0.95 (3H, t, J = 7 Hz, C-17), 1.02 (3H, d, J = 7 Hz, C-21), 1.05 (1H, ddd, $J = 4$, 12, 16 Hz, C-7), 1.23 (3H, d, $J = 7$ Hz, C-18), 1.60-1.73 (3H, m, C-7(1H), C-16(2H)), 1.55 (lH, dq, J = 7, 10 Hz, C-4), 1.80-1.95 (lH, m, C-19), 1.80 (3H, s, C-22), 1.94 (lH, dd, J $= 2$, 17 Hz, C-2), 2.01-2.07 (1H, m, C-6), 2.19 (0.23H, dd, J = 6, 13 Hz, C-19), 2.27 (0.77H, dd, J = 6, 13 Hz, C-19), 2.45-2.65 (lH, m, C-8), 2.57 (lH, dd, J = 11, 17 Hz, C-2), 2.88 (lH, ddt, $J = 5$, 6, 10 Hz, C-14), 3.67 (1H, dd, $J = 2$, 11 Hz, C-3), 3.68-3.77 (2H, m, C-23), 3.80 $(0.23H, dd, J = 4, 10 Hz, C-5), 4.12 (0.77H, dd, J = 4, 10 Hz, C-5), 4.93 (1H, dt, J = 2, 10$ Hz, C-15), 5.42 (0.23H, d, J = 6 Hz, C-20), 5.50 (0.77H, dd, J = 4, 6 Hz, C-20), 5.80 (lH, d, J $= 10$ Hz, C-13), 6.33 (0.77H, d, J = 15 Hz, C-10), 6.37 (0.23H, d, J = 15 Hz, C-10), 7.22 (1H, d, $J = 15$ Hz, C-11). 20: δ (270 MHz) 0.84 (3H, d, J = 7.0 Hz), 0.86 (3H, t, J = 7.5 Hz), 1.06 (3H, d, J = 7.0 Hz), 1.10-2.00 (8H, m), 1.36 (3H, s), 1.41 (3H, s), 1.78 (3H, s), 2.21 (lH, dd, $J = 15.0, 4.0 \text{ Hz}$, 2.51 (1H, dd, $J = 15.0, 9.0 \text{ Hz}$), 2.85-3.25 (2H, m), 3.13 (2H, d, $J = 22.0$ Hz), 3.38-3.60 (4H, m), 3.63-4.05 (2H, m), 3.74 (3H, d, J = 1.0 Hz), 3.80 (6H, s), 3.87 (3H, s), 3.88 (3H, s), 4.32-4.45 (4H, m), 5.21 (lH, m), 6.45 (lH, d, J = 10.0 Hz), 6.80-6.95 (5H, m), 7.20 (2H, d, J = 9.0 Hz), 9.44 (lH, s). 21: 6 0.83 (3H, d, J = 6.5 Hz), 0.88 (3H, t, J = 7.0 Hz), 1.10-2.20 (8H, m), 1.13 (3H, d, J = 6.5 Hz), 1.38 (3H, s), 1.43 (3H, s), 1.82 (3H, s), 2.34 (2H, t, J = 5.0 Hz), 2.80-3.10 (3H, m), 3.40 (2H, t, J = 6.5 Hz), 3.46 (2H, d, J = 5.0 Hz), 3.81 (3H, s), 3.88 (3H, s), 3.90 (3H, s), 3.90-4.20 (2H, m), 4.35-4.50 (4H, m), 4.99 (lH, dt, $J = 4.0$, 7.5 Hz), 5.81 (1H, d, $J = 11.0$ Hz), 6.17 (1H, d, $J = 16.0$ Hz), 6.80-6.95 (5H, m), 7.17 (1H, d, J = 16.0 Hz), 7.20 (2H, d, J = 9.0 Hz). m/z (FD) 736 (M⁺, 100%).

(Received in Japan 8 May 1986)